SAFETY DATA SHEET B-Cap[™] 27 Antimicrobial Agent

SDS #: 7722-84-1-27-80 Revision date: 2015-05-08 Format: NA Version 1



1. PRODUCT AND COMPANY IDENTIFICATION

Product Identifier			
Product Name	B-Cap™ 27 Antimicrobial Agent		
Other means of identification			
EPA Registration Number	72372-4		
CAS-No Synonyms	7722-84-1 Hydrogen peroxide		
Recommended use of the chemical	and restrictions on use		
Recommended Use:	Antimicrobial agent used for biofouling control in cooling and process water systems, as a paper mill biocide, and for general slime control; PeroxyChem acquired the EPA product registration from FMC		
Restrictions on Use:	Use as recommended by the label.		
<u>Manufacturer/Supplier</u>	PeroxyChem LLC 2005 Market Street Suite 3200 Philadelphia, PA 19103 Phone: +1 267/ 422-2400 (General Information) E-Mail: sdsinfo@peroxychem.com For leak, fire, spill or accident emergencies, call: 1 800 / 424 9300 (CHEMTREC - U.S.A.) 1 703 / 527 3887 (CHEMTREC - Collect - All Other Countries) 1 303/ 389-1409 (Medical - U.S Call Collect) 1 281 / 474-8750 (Bayport, Texas Plant)		

2. HAZARDS IDENTIFICATION

Classification

OSHA Regulatory Status

This material is considered hazardous by the OSHA Hazard Communication Standard (29 CFR 1910.1200).

Acute toxicity - Oral	Category 4
Serious eye damage/eye irritation	Category 1
Oxidizing Liquids	Category 2

GHS Label elements, including precautionary statements

EMERGENCY OVERVIEW

Danger

Hazard Statements

H318 - Causes serious eye damage H302 - Harmful if swallowed

H272 - May intensify fire; oxidizer



Precautionary Statements - Prevention

P264 - Wash face, hands and any exposed skin thoroughly after handling

P280 - Wear protective gloves/ protective clothing/ eye protection/ face protection

P210 - Keep away from heat/sparks/open flames/hot surfaces. - No smoking

P220 - Keep/Store away from clothing/flammable materials/combustibles

P221 - Take any precaution to avoid mixing with combustibles/flammables

Precautionary Statements - Response

P305 + P351 + P338 - IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing

P310 - Immediately call a POISON CENTER or doctor

P301 + P312 - IF SWALLOWED: Call a POISON CENTER or doctor if you feel unwell

P330 - Rinse mouth

P370 + P378 - In case of fire: Use water for extinction

Hazards not otherwise classified (HNOC)

No hazards not otherwise classified were identified.

Other Information

IF INHALED: Remove person to fresh air and keep comfortable for breathing. Get medical attention if symptoms occur. IF ON SKIN OR CLOTHING: Wash with plenty of water. Take off contaminated clothing and wash before reuse. If skin irritation occurs: Get medical advice/attention.

Keep container in a cool place out of direct sunlight. Store only in vented containers. Do not store on wooden pallets. Do not return unused material to its original container. Avoid contamination - Contamination could cause decomposition and generation of oxygen which may result in high pressure and possible container rupture. Empty drums should be triple rinsed with water before discarding.

3. COMPOSITION/INFORMATION ON INGREDIENTS

Formula

HO - OH

Chemical name	CAS-No	Weight %
Hydrogen peroxide	7722-84-1	27
Water	7732-18-5	73

Occupational exposure limits, if available, are listed in section 8

4. FIRST AID MEASURES

	Version 1
	Remove contact lenses, if present, after the first 5 minutes, then continue rinsing. Seek immediate medical attention/advice.
Skin Contact	Take off contaminated clothing. Rinse skin immediately with plenty of water for 15-20 minutes. Call a poison control center or doctor for further treatment advice.
Inhalation	Move to fresh air. If person is not breathing, contact emergency medical services, then give artificial respiration, preferably mouth-to-mouth if possible. Call a poison control center or doctor for further treatment advice.
Ingestion	Rinse mouth. Do not induce vomiting. If conscious, give 2 glasses of water. Get immediate medical attention. Never give anything by mouth to an unconscious person.
Most important symptoms and effects, both acute and delayed	Hydrogen Peroxide irritates respiratory system and, if inhaled, may cause inflammation and pulmonary edema. The effects may not be immediate. Overexposure symptoms are coughing, giddiness and sore throat. In case of accidental ingestion, necrosis may result from mucous membrane burns (mouth, esophagus and stomach). Oxygen rapid release may cause stomach swelling and hemorrhaging, which may product major, or even fatal, injury to organs if a large amount has been ingested. In case of skin contact, may cause burns, erythema, blisters or even necrosis.
Indication of immediate medical attention and special treatment needed, if necessary	Hydrogen peroxide at these concentrations is a strong oxidant. Direct contact with the eye is likely to cause corneal damage especially if not washed immediately. Careful opthalmologic evaluation is recommended and the possibility of local corticosteroid therapy should be considered. Because of the likelihood of corrosive effects on the gastrointestinal tract after ingestion, and the unlikelihood of systemic effects, attemps at evacuating the stomach via emesis induction or gastric lavage should be avoided. There is a remote possibility, however, that a nasogastric or orogastric tube may be required for the reduction of severe distension due to gas formation.
	5. FIRE-FIGHTING MEASURES
Suitable Extinguishing Media	S. FIRE-FIGHTING MEASURES Water. Do not use any other substance.
Suitable Extinguishing Media Specific Hazards Arising from the Chemical	
Specific Hazards Arising from the	Water. Do not use any other substance. In closed unventilated containers, risk of rupture due to the increased pressure from
Specific Hazards Arising from the Chemical	Water. Do not use any other substance. In closed unventilated containers, risk of rupture due to the increased pressure from decomposition. Contact with combustible material may cause fire
Specific Hazards Arising from the Chemical Hazardous Combustion Products <u>Explosion data</u> Sensitivity to Mechanical Impact	Water. Do not use any other substance. In closed unventilated containers, risk of rupture due to the increased pressure from decomposition. Contact with combustible material may cause fire On decomposition product releases oxygen which may intensify fire. Not sensitive.
Specific Hazards Arising from the Chemical Hazardous Combustion Products <u>Explosion data</u> Sensitivity to Mechanical Impact Sensitivity to Static Discharge Protective equipment and	 Water. Do not use any other substance. In closed unventilated containers, risk of rupture due to the increased pressure from decomposition. Contact with combustible material may cause fire On decomposition product releases oxygen which may intensify fire. Not sensitive. Not sensitive. Use water spray to cool fire exposed surfaces and protect personnel. Move containers from fire area if you can do it without risk. As in any fire, wear self-contained breathing apparatus
Specific Hazards Arising from the Chemical Hazardous Combustion Products <u>Explosion data</u> Sensitivity to Mechanical Impact Sensitivity to Static Discharge Protective equipment and	 Water. Do not use any other substance. In closed unventilated containers, risk of rupture due to the increased pressure from decomposition. Contact with combustible material may cause fire On decomposition product releases oxygen which may intensify fire. Not sensitive. Not sensitive. Use water spray to cool fire exposed surfaces and protect personnel. Move containers from fire area if you can do it without risk. As in any fire, wear self-contained breathing apparatus and full protective gear.
Specific Hazards Arising from the Chemical Hazardous Combustion Products Explosion data Sensitivity to Mechanical Impact Sensitivity to Static Discharge Protective equipment and precautions for firefighters	 Water. Do not use any other substance. In closed unventilated containers, risk of rupture due to the increased pressure from decomposition. Contact with combustible material may cause fire On decomposition product releases oxygen which may intensify fire. Not sensitive. Not sensitive. Use water spray to cool fire exposed surfaces and protect personnel. Move containers from fire area if you can do it without risk. As in any fire, wear self-contained breathing apparatus and full protective gear. 6. ACCIDENTAL RELEASE MEASURES Avoid contact with skin, eyes and clothing. Wear personal protective equipment. Isolate and post spill area. Keep people away from and upwind of spill/leak. Eliminate all sources of

Methods for Containment Dike to collect large liquid spills. Stop leak and contain spill if this can be done safely. Small

spillage: Dilute with large quantities of water.

Methods for cleaning up

Flush area with flooding quantities of water. Hydrogen peroxide may be decomposed by adding sodium metabisulfite or sodium sulfite after diluting to about 5%.

7. HANDLING AND STORAGE

Handling	Keep/Store away from clothing/ combustible materials. Wear personal protective equipment. Reference to other sections. Never return unused hydrogen peroxide to original container. Contamination may cause decomposition and generation of oxygen gas which could result in high pressures and possible container rupture. Empty drums should be triple rinsed with water before discarding. Utensils used for handling hydrogen peroxide should only be made of glass, stainless steel, aluminum or plastic. Pipes and equipment should be passivated before first use. Use only in well-ventilated areas. Hydrogen peroxide should be stored only in vented containers and transferred only in a prescribed manner.
Storage	Keep containers in cool areas out of direct sunlight and away from combustibles. Provide mechanical general and/or local exhaust ventilation to prevent release of vapor or mist into work environment. Containers must be vented. Keep/store only in original container. Store rooms or warehouses should be made of non-combustible materials with impermeable floors. In case of release, spillage should flow to safe area. Containers should be visually inspected on a regular basis to detect any abnormalities (swollen drums, increases in temperature, etc.).
Incompatible products	Combustible materials. Copper alloys, galvanized iron. Strong reducing agents. Heavy metals. Iron. Copper alloys. Contact with metals, metallic ions, alkalis, reducing agents and organic matter (such as alcohols or terpenes) may produce self-accelerated thermal decomposition.

8. EXPOSURE CONTROLS/PERSONAL PROTECTION

Control parameters

Exposure Guidelines

Ingredients with workplace control parameters.

Chemical name	ACGIH TLV	OSHA PEL	NIOSH	Mexico
Hydrogen peroxide 7722-84-1	TWA: 1 ppm	TWA: 1 ppm TWA: 1.4 mg/m ³	IDLH: 75 ppm TWA: 1 ppm TWA: 1.4 mg/m ³	Mexico: TWA 1 ppm Mexico: TWA 1.5 mg/m ³ Mexico: STEL 2 ppm Mexico: STEL 3 mg/m ³
Chemical name	British Columbia	Quebec	Ontario TWAEV	Alberta
Hydrogen peroxide 7722-84-1	TWA: 1 ppm	TWA: 1 ppm TWA: 1.4 mg/m ³	TWA: 1 ppm	TWA: 1 ppm TWA: 1.4 mg/m ³

Appropriate engineering controls

Engineering measures Ensure that eyewash stations and safety showers are close to the workstation location. Ensure adequate ventilation.

Individual protection measures, such as personal protective equipment

- Eye/Face ProtectionUse chemical splash-type monogoggles and a full-face shield made of polycarbonate,
acetate, polycarbonate/acetate, PETG or thermoplastic.
- Skin and Body Protection For body protection wear impervious clothing such as an approved splash protective suit made of SBR rubber, PVC (PVC Outershell w/Polyester Substrate), Gore-Tex (Polyester trilaminate w/Gore-Tex), or a specialized HAZMAT Splash or Protective Suite (Level A, B, or C). For foot protection, wear approved boots made of NBR, PVC, Polyurethane, or neoprene. Overboots made of Latex or PVC, as well as firefighter boots or specialized HAZMAT boots are also permitted. DO NOT wear any form of boot or overboot made of nylon or nylon blends. DO NOT USE cotton, wool or leather as these materials react rapidly with higher concentrations of hydrogen peroxide. Completely submerge hydrogen peroxide

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contaminated clothing or other materials in water prior to drying. Residual hydrogen peroxide, if allowed to dry on materials such as paper, fabrics, cotton, leather, wood or other combustibles, can cause the material to ignite and result in a fire.

Hand ProtectionFor hand protection, wear approved gloves made of nitrile, PVC, or neoprene. DO NOT
use cotton, wool or leather for these materials react RAPIDLY with higher concentrations of
hydrogen peroxide. Thoroughly rinse the outside of gloves with water prior to removal.
Inspect regularly for leaks.Respiratory ProtectionIf concentrations in excess of 10 ppm are expected, use NIOSH/DHHS approved
self-contained breathing apparatus (SCBA) or other approved air-supplied respirator (ASR)
equipment (e.g., a full-face airline respirator (ALR)). DO NOT use any form of air-purifying
respirator (APR) or filtering facepiece (dust mask), especially those containing oxidizable

Hygiene measures Avoid breathing vapors, mist or gas. Clean water should be available for washing in case of eye or skin contamination.

sorbants such as activated carbon.

9. PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Appearance Physical State Color Odor Odor threshold pH Melting point/freezing point Boiling Point/Range Flash point Evaporation Rate Flammability (solid, gas) Flammability Limit in Air Upper flammability limit: Lower flammability limit:	Clear, colorless liquid Liquid Colorless odorless Not applicable <= 3.7 -22 °C 105 °C Not flammable No information available Not flammable Not applicable
Vapor pressure	26 mm Hg @ 30 °C
Vapor density	No information available
Density	1.10 g/cm ³ @ 20°C
Specific gravity	1.10
Water solubility	completely soluble
Solubility in other solvents	No information available
Partition coefficient	log Kow = -1.5 @ 20 °C
Autoignition temperature	Not combustible
Decomposition temperature	100 °C
Viscosity, kinematic	1.06 cP @ 20 °C
Viscosity, dynamic	No information available
Explosive properties	No information available
Oxidizing properties	Strong oxidizer
Molecular weight	34
Bulk density	Not applicable

10. STABILITY AND REACTIVITY

Reactivity

Chemical Stability

Stable under normal conditions. Decomposes on heating. Stable under recommended storage conditions.

Possibility of Hazardous Reactions Contact with organic substances may cause fire or explosion. Contact with metals, metallic

General information Protective engineering solutions should be implemented and in use before personal protective equipment is considered.

ions, alkalis, reducing agents and organic matter (such as alcohols or terpenes) may produce self-accelerated thermal decomposition.

Hazardous polymerization	Hazardous polymerization does not occur.
Conditions to avoid	Excessive heat; Contamination; Exposure to UV-rays; pH variations.
Incompatible materials	Combustible materials. Copper alloys, galvanized iron. Strong reducing agents. Heavy metals. Iron. Copper alloys. Contact with metals, metallic ions, alkalis, reducing agents and organic matter (such as alcohols or terpenes) may produce self-accelerated thermal decomposition.

Hazardous Decomposition Products Oxygen which supports combustion. Liable to produce overpressure in container.

11. TOXICOLOGICAL INFORMATION

Product Information

LD50 Oral	50% solution: LD50 > 225 mg/kg bw (rat) 35 % solution:LD50 1193 mg/kg bw (rat) 70 % solution: LD50 1026 mg/kg bw (rat)
LD50 Dermal	35% solution: LD50 > 2000 mg/kg bw (rabbit)
LC50 Inhalation	70 % solution: LD50 9200 mg/kg bw (rabbit) 50% solution: LC50 > mg/m ³ (rat) (4-hr) Hydrogen Peroxide vapors: LC0 9400 mg/m ³ (mouse) (5 - 15 minutes) Hydrogen Peroxide vapors: LC50 > 2160 mg/m ³ (mouse)
Serious eye damage/eye irritation Skin corrosion/irritation	Corrosive. Severely irritating to the eyes. Moderately irritating (rabbit).
Sensitization	Did not cause sensitization on laboratory animals.

Information on toxicological effects

Symptoms

Inhalation of corrosive fumes/gases may cause coughing, choking, headache, dizziness, and weakness for several hours. Pulmonary edema may occur with tightness in the chest, shortness of breath, bluish skin, decreased blood pressure, and increased heart rate.

Delayed and immediate effects as well as chronic effects from short and long-term exposure

CarcinogenicityThis product contains hydrogen peroxide. The International Agency for Research on
Cancer (IARC) has conculded that there is inadequate evidence for carcinogenicity of
hydrogen peroxide in humans, but limited evidence in experimental animals (Group 3 - not
classifiable as to its carcinogenicity to humans). The American Conference of
Governmental Industrial Hygienists (ACGIH) has concluded that hydrogen peroxide is a
'Confirmed Animal Carcinogen with Unknown Relevance to Humans' (A3).

Chemical name	ACGIH	IARC	NTP	OSHA
Hydrogen peroxide	A3	3		
7722-84-1				
Mutagenicity		s not recognized as muta lid not show mutagenic ef	genic by Research Agenci fects	es
Reproductive toxicity	This product is not recognized as reprotox by Research Agencies. No toxicity to reproduction in animal studies.			
STOT - single exposure	Not classified			
STOT - repeated exposure	Not classified			

Target organ effects

Eyes, Respiratory System, Skin.

No information available.

Aspiration hazard

12. ECOLOGICAL INFORMATION

Ecotoxicity

Ecotoxicity effects

Hydrogen peroxide is naturally produced by sunlight (between 0.1 and 4 ppb in air and 0.001 to 0.1 mg/L in water). Not expected to have significant environmental effects.

Hydrogen peroxide (772	2-84-1)			
Active Ingredient(s)	Duration	Species	Value	Units
Hydrogen peroxide	96 h LC50	Fish Pimephales promelas	16.4	mg/L
Hydrogen peroxide	72 h LC50	Fish Leuciscus idus	35	mg/L
Hydrogen peroxide	48 h EC50	Daphnia pulex	2.4	mg/L
Hydrogen peroxide	24 h EC50	Daphnia magna	7.7	mg/L
Hydrogen peroxide	72 h EC50	Algae Skeletonema costatum	1.38	mg/L
Hydrogen peroxide	21 d NOEC	Daphnia magna	0.63	mg/L

Persistence and degradability	Hydrogen peroxide in the aquatic environment is subject to various reduction or oxidation processes and decomposes into water and oxygen. Hydrogen peroxide half-life in freshwater ranged from 8 hours to 20 days, in air from 10 - 20 hours, and in soils from minutes to hours depending upon microbiological activity and metal contamination.	
Bioaccumulation	Material may have some potential to bioaccumulate but will likely degrade in most environments before accumulation can occur.	
Mobility	Will likely be mobile in the environment due to its water solubility but will likely degrade over time.	
Other Adverse Effects	Decomposes into oxygen and water. No adverse effects.	
	13. DISPOSAL CONSIDERATIONS	
Waste disposal methods	Dispose of in accordance with local regulations. Can be disposed as waste water, when in compliance with local regulations.	
US EPA Waste Number	D001	
Contaminated Packaging	Dispose of in accordance with local regulations. Drums - Empty as thoroughly as possible. Triple rinse drums before disposal. Avoid contamination; impurities accelerate decomposition. Never return product to original container.	

14. TRANSPORT INFORMATION

DOT

UN/ID no	2014
Proper Shipping Name	HYDROGEN PEROXIDE, AQUEOUS SOLUTION
Hazard class	5.1
Subsidiary class	8
Packing Group	II
<u>TDG</u> UN/ID no	UN 2014

Proper Shipping Name Hazard class Subsidiary class Packing Group	HYDROGEN PEROXIDE, AQUEOUS SOLUTION 5.1 8 II
ICAO/IATA	Air regulation permit shipment of Hydrogen Peroxide (<=40%) in non-vented containers for Air Cargo Only aircraft, as well as for Passenger and Cargo aircraft. HOWEVER, all PeroxyChem Hydrogen Peroxide containers are vented and therefore, air shipments of PeroxyChem H2O2 are not permitted. IATA air regulations state that venting of packages containing oxidizing substances is not permitted for air transport.
IMDG/IMO UN/ID no Proper Shipping Name Hazard class Subsidiary Hazard Class Packing Group	UN 2014 HYDROGEN PEROXIDE, AQUEOUS SOLUTION 5.1 8 II
OTHER INFORMATION	Protect from physical damage. Keep drums in upright position. Drums should not be stacked in transit. Do not store drums on wooden pallets.

15. REGULATORY INFORMATION

U.S. Federal Regulations

SARA 313

Section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 (SARA). This product does not contain any chemicals which are subject to the reporting requirements of the Act and Title 40 of the Code of Federal Regulations, Part 372

SARA	311/312	Hazard	Categories

Acute health hazard	Yes
Chronic health hazard	No
Fire hazard	Yes
Sudden release of pressure hazard	No
Reactive Hazard	No

Clean Water Act

This product does not contain any substances regulated as pollutants pursuant to the Clean Water Act (40 CFR 122.21 and 40 CFR 122.42)

CERCLA

Chemical name	Hazardous Substances RQs	Extremely Hazardous Substances RQs	SARA RQ
Hydrogen peroxide 7722-84-1		1000 lb	

Hydrogen Peroxide RQ is for concentrations of > 52% only

FIFRA INFORMATION

EPA Pesticide registration number 72372-4

This chemical is a pesticide product registered by the Environmental Protection Agency and is subject to certain labeling requirements under federal pesticide law. These requirements differ from the classification criteria and hazard information required for safety data sheets, and for workplace labels of non-pesticide chemicals. Following is the hazard information as required on the pesticide label:

DANGER

Corrosive. Causes irreversible eye damage and skin burns. MAY BE FATAL IF INHALED

HARMFUL IF SWALLOWED OR ABSORBED THROUGH SKIN Strong oxidizing agent. This pesticide is toxic to birds, mammals, fish and aquatic invertebrates.

International Inventories

Component	TSCA (United States)	DSL (Canada)	EINECS/EL INCS (Europe)	ENCS (Japan)	China (IECSC)	KECL (Korea)	PICCS (Philippines)	AICS (Australia)	NZIoC (New Zealand)
Hydrogen peroxide 7722-84-1 (27)	Х	X	Х	Х	Х	Х	X	х	х

Mexico - Grade

Serious risk, Grade 3

CANADA

WHMIS Hazard Class

C - Oxidizing materials D2B - Toxic materials E - Corrosive material







16. OTHER INFORMATION

NFPA	Health Hazards 3	Flammability 0	Stability 1	Special Hazards OX		
HMIS	Health Hazards 3	Flammability 0	Physical hazard 1	Special precautions H		
NFPA/HMIS Ratings Leg	Special Haz Protection =	Severe = 4; Serious = 3; Moderate = 2; Slight = 1; Minimal = 0 Special Hazards: OX = Oxidizer Protection = H (Safety goggles, gloves, apron, the use of supplied air or SCBA respirator is required in lieu of a vapor cartidge respirator)				
Uniform Fire Code	Oxidizer: Cla	ass 1Liquid				
Revision date: Revision note	2015-05-08 Initial Releas	Se				

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Prepared By:

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